St. Paul's Convent School Dyeing Hair, Dying Hair? (「髮」證先鋒)

Introduction

Hair dyeing is the norm in this era but are we letting our hair turn into dying hair in the process?

Objectives

In Part I, the content of bleaching agent H_2O_2 of 9 commercial hair dyes was determined through titration. In Part IIA, we investigated the chemical, temperature and time effect on hair bleaching. In Part IIB, we compared the damaging effects of different bleaching conditions on hair. In Part IIIA, we used lemon juice with tea to bleach hair. In Part IIIB, we compared the dyeing effects of commercial and natural dyes. In Part IV, we investigated the effect of

the 7 natural dyes on white hair.

Chemical Principles

Part I Determination of H₂O₂ Content in Commercial Hair Dyes

The content of active ingredient, H₂O₂, was determined through titration against acidified KMnO₄.

 $2 \text{ MnO}_{4}^{-}_{(aq)} + 5 \text{ H}_2\text{O}_{2}_{(aq)} + 6 \text{ H}^+_{(aq)} \rightarrow 2 \text{ Mn}^{2+}_{(aq)} + 5 \text{ O}_{2}_{(g)} + 8 \text{ H}_2\text{O}_{(l)}$

Under heating, the rate of decomposition of H_2O_2 increases.

 $2 \; H_2 O_{2 \; (aq)} \; \rightarrow \; 2 H_2 O_{\; (l)} + O_{2 \; (g)}$

Part IIA Chemical, Temperature and Time Effect on Hair Bleaching

Hair bleaching involves the dissolution of melanin pigment granules and subsequent depolymerisation of pigment chromophore into carboxylated derivatives which are soluble in alkaline medium and can be eliminated by rinsing. A typical bleaching includes a bleaching agent, usually H_2O_2 , a NH₃ base, which raises the cuticle scales for the penetration of bleaching chemicals and a booster like $S_2O_8^{2^2}$, which provides an additional source of oxygen for bleaching.

Part IIB Damaging Effects of Different Bleaching Conditions

During bleaching, H_2O_2 attacks keratin protein in the hair cortex by oxidizing the amino acid cystine to cysteic acid and breaking the disulphide linkages. As bleaching compositions are usually formulated between pH 9 to 11, the hydrolysis of peptide bond during bleaching could

ascorbic acid



Part IIIA Hair Bleaching with Lemon Juice and Tea

also reduce the tensile strength of hair.

Ascorbic acid in lemon juice and catechins in tea are antioxidants which convert the double bonds of the conjugated π system of melanin chromophoric units into single bonds, thus the chromophore is unable to absorb visible light and hair is said to be bleached. Sunlight is required for significant bleaching results as it opens the cuticle by altering the amino acids by photochemical degradation.

Part IIIB Comparison of Dyeing Effects of Commercial and Natural Dyes

Permanent commercial hair dyes generally consist of p-diamines and p-aminophenols that are oxidized by H₂O₂ to active intermediates, which react inside the hair with colour couplers to provide shampoo-resistant hair dyes. The natural sources chosen cover a wide range of colours and all contained high concentrations of colour pigments, which are anthocyanins, chlorophyll, curcumin, beta-carotene and caffetannin respectively. Addition of binding agents like egg yolk enhances the molecular interactions between keratin and pigments. Egg yolk and soya flour





raise hair cuticle



contains lecithin, a phospholipid which increases the adhesion of natural dyes to the hair. Egg white is an albumin solution which gives shine and transparency to colours. Gelatin has an amphoteric and hydrophobic character which gives it emulsifying properties.

10-point scale

Experiments

Part I Determination of H₂O₂ Content in Commercial Hair Dyes

Known volumes of H₂O₂ developers were acidified and titrated against standard KMnO₄ solution until a permanent pink colour was observed. Titration was carried out at different temperature to investigate the effect of temperature on the H_2O_2 content in the developers.

Part IIA Chemical, Temperature and Time Effect on Hair Bleaching

All the black hair used is from a 12 year old girl whose hair has never received any treatment. Hair bleached with different combinations of bleaching agent, base and booster for 1 hour, 3 hours, 24 hours and at 25°C, 40 °C and 60 °C, were rated according to a ten point scale.

Part IIB Damaging Effects of Different Bleaching Conditions

A spring balance was used to measure the average force required to snap a piece of hair taped to the tabletop in which a larger force implied a stronger tensile strength of the hair.

Part IIIA Hair Bleaching with Lemon Juice and Tea

Lemon juice was freshly squeezed and added to 11 different types of tea in the volume ratio 1:3 for 24 hours under sunlight to observe the bleaching effects.

Part IIIB Comparison of Dyeing Effects of Commercial and Natural Dyes Pigments were extracted from 7 natural sources: strawberries, blueberries, cherries, spinach, ginger, carrot and coffee for hair dyeing. Ripen fruits were squashed with a glass rod in a beaker of heated vinegar and water mixed in the

volume ratio of 1:7. Suction filtration was then carried out to obtain the dye extract. Part of the hair samples was dipped into one of the 4 binding agents before dyeing and dyeing effect was observed for commercially and naturally dyed hair. Hair samples were washed

with shampoo to test for the permanence of the natural dyes.

Part IV Effect of Natural Dyes on White Hair

White hair samples were taken from a 65 years old lady. Hair was bleached with bleaching cream and 6% H₂O₂ developer at 40°C for 1 hour, 1.5 hours and 2 hours and then dyed with blueberry dye extract for 3 hours, 6 hours, 9 hours and 12 hours respectively to investigate the effect of duration of bleaching and dyeing on hair.















bleaching cream. 6% H₂O₂ developer





greatest bleaching power



Discussion

In Part I, Garnier – Nutrisse had the highest H₂O₂ content, followed by Wella and Garnier. The variation of H₂O₂ content against temperature increase was found to be insignificant.

Garnier-Nutrisse

In Part IIA, bleaching effect was significantly increased with the addition of a bleaching base and the

presence of a booster could further enhance the bleaching effect. Without a base, only limited bleaching effect can be observed even at 60°C. The concluded best bleaching condition was using bleaching cream and 6% H₂O₂ developer for either 1 hour at 40°C or 3 hours at 25°C. In Part IIB, hair bleached by Wella had the highest tensile strength. Tensile strengths for hair bleached at 25°C and 40 °C were similar but that for 60°C declined greatly. Wella

Garnier-Nutrisse

In Part IIIA, lemon juice together with green tea produced a better bleaching effect while white and black tea almost had no bleaching

effect. In Part IIIB, binding agents was found to improve the dyeing effect and tensile strength of naturally dyed hair samples. Cherry dyed hair produced a similar reddish brown colour as Wella and Garnier - Nutrisse, which were the most effective dyeing brands. Colour of naturally dyed hair remained even after 5 times of washing. Thus, first bleaching black hair with bleaching cream and 6% H₂O₂ developer at 40 °C for 1 hour, then dyeing with egg white Number of

0

washings

Colour

1

2

and cherry extract for 3 hours was recommended for natural dyeing of reddish brown hair.

In Part IV, white hair bleached for 1 hour and dyed for 3 hours produced the best black dyeing effect. It was recommended bleach their white hair by following the previous procedure, then dyeing with egg white and blueberry or black bean extract for 3 hours.

Conclusion



Natural dyes are effective, cheaper and much less damaging in hair dyeing than commercial dyes. Further research could definitely make natural dyeing methods more convenient and feasible.





	677

100	150
	- 50

3

4

5

egg white blueberry extract